# RESEARCH



# Dissolved inorganic carbon input significantly lowers carbon dioxide flux but not methane flux in shallow macrophyte-dominated systems with positive effects on carbon stocks

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# Abstract

**Background** With the increase in the inorganic carbon input from watersheds, elevated dissolved inorganic carbon (DIC) concentrations will significantly impact the carbon cycle in freshwater ecosystems. Moreover, the limited diffusion rate of  $CO_2$  in water, coupled with the lack of functional stomata, greatly restricts the ability of submerged macrophytes to absorb  $CO_2$  from their aquatic environment. The importance of bicarbonate (HCO<sub>3</sub><sup>-</sup>) for submerged macrophytes becomes more pronounced. Current research focuses on the effects of DIC (notably HCO<sub>3</sub><sup>-</sup>) on the phenotypic plasticity of submerged macrophytes, while its impact on their carbon stock capabilities has rarely been reported.

**Results** In this study, *Myriophyllum spicatum* served as the model macrophyte within a mesocosm experimental system to assess the impact of  $HCO_3^-$  enrichment (0.5 to 2.5 mmol L<sup>-1</sup>) on carbon stocks and emissions across a one-year period. Our findings indicated that the addition of  $HCO_3^-$  had a non-significant inhibitory effect on the diffusive fluxes of methane (CH<sub>4</sub>) emissions. Concurrently, it significantly reduced CO<sub>2</sub> fluxes within the systems. The annual average CO<sub>2</sub> fluxes across the four  $HCO_3^-$  addition levels were  $-3.48 \pm 7.60$ ,  $-6.78 \pm 5.87$ ,  $-7.15 \pm 8.68$ , and  $-14.04 \pm 14.39$  mol m<sup>-2</sup> yr<sup>-1</sup>, respectively, showing significant differences between low /medium- and high-  $HCO_3^-$  addition levels.

**Conclusion** The addition of HCO<sub>3</sub><sup>-</sup> enhanced carbon stocks in water, macrophytes and the entire system, with minimal effects on carbon sedimentation stocks. Our study provides valuable insights into understanding the carbon sink capacity of aquatic ecosystems and elucidates the underlying mechanisms driving these processes on a system scale.

**Keywords** Carbon stock, Greenhouse gas flux, Dissolved inorganic carbon, Submerged macrophyte, Carbon sedimentation

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# Introduction

Greenhouse gas (GHG) emissions from lakes constitute a crucial part of the global carbon budget. The annual emissions of CO2 and CH4 from global lakes are estimated at approximately 0.53 Pg C and 0.15 Pg C, respectively [44]. The carbon cycle within lakes significantly influences their GHG emissions. Carbon cycling encompasses various processes: the fixation of inorganic carbon by primary producers through photosynthesis, the conversion into various forms of organic carbon (OC), carbon emissions, and carbon sedimentation [63]. The effects of carbon decomposition are ultimately reflected in changes in organic or inorganic carbon concentrations in systems and CO<sub>2</sub> and CH<sub>4</sub> emissions [57, 59]. Lake carbon cycling can be greatly affected by external carbon inputs. It is estimated that at least  $1.13 \pm 0.33$  billion tons of inorganic carbon, predominantly in the form of DIC, are annually lost to inland waters through soil erosion [20]. With the increase in inorganic carbon inputs, DIC has become increasingly significant for the photosynthetic carbon fixation of submerged macrophytes.

In the process of photosynthetic carbon fixation by submerged macrophytes,  $CO_2$  is the most easily acquired form of inorganic carbon for it can enter the cells through passive transport [41]. Nonetheless, the diffusion rate of  $CO_2$  in water is only one ten-thousandth of that in air. Submerged macrophytes lack functional stomata, which limited their ability to acquire  $CO_2$  from water [37]. To address the stress of limited  $CO_2$  availability in aquatic environments, submerged macrophytes have evolved a strategy to utilize bicarbonate ( $HCO_3^-$ ) as an alternative carbon source for photosynthesis [39]. The comprehension of how DIC input influences the balance among carbon fixation, sedimentation, and emissions is crucial for predicting carbon stocks under global changes.

The impact of submerged macrophytes on  $CO_2$  and CH<sub>4</sub> emissions is still under debate. DIC can promote the growth of macrophytes and enhance the conversion to higher levels of biological carbon bound in their biomass [34, 39]. These biological carbon compounds are often recalcitrant compounds (e.g., cellulose and lignin), thus they easily form refractory carbon pools and thereby increasing carbon stock capacity [26, 54]. However, the increase in submerged macrophyte biomass also implies a rise in CO<sub>2</sub> and CH<sub>4</sub> produced through their respiration and decomposition after decay [12]. Additionally, the increase in submerged macrophyte biomass may directly or indirectly promote CH<sub>4</sub> formation via aerobic methanogenesis or co-metabolic effects [1, 42]. Submerged macrophytes might directly contribute to CH<sub>4</sub> formation by generating strong oxidants that drive methyl radical production to form CH<sub>4</sub> [17]. Some perspectives suggested that submerged macrophytes had an insignificant impact on lake  $CO_2$  and  $CH_4$  emissions, regardless of short-term effects over three months [14, 16] or longterm effects over one year [2]. However, other perspectives argued that an increase in submerged macrophyte biomass can enhance  $CH_4$  emissions while reducing  $CO_2$ emissions [58]. The impact of submerged macrophytes on  $CO_2$  and  $CH_4$  emissions, as well as the mechanisms driving such effects, remains unexplored in the context of external DIC input. Thus, a detailed understanding of how  $HCO_3^-$  affects carbon cycling is important to predict how DIC will influence carbon stocks in lakes.

To date, few comprehensive studies have explored the effects of DIC, particularly HCO3<sup>-</sup>, on carbon stocks and emissions in submerged macrophyte-dominated systems. To address this research gap, we conducted a year-long 450 L mesocosm experiment to investigate the influence of HCO<sub>3</sub><sup>-</sup> on carbon stocks as well as CO<sub>2</sub> and CH<sub>4</sub> fluxes. The experiment employed *Myriophyllum spi*catum (M. spicatum), a submerged, rooted freshwater plant species widely distributed across Europe, Asia, and North America, renowned for its efficient utilization of  $HCO_3^{-}$  [62]. The mesocosms were subjected to subtropical (China) conditions with three different HCO<sub>3</sub><sup>-</sup> addition scenarios (0.5, 1.0, and 2.5 mmol  $L^{-1}$ ). During the experiment, we measured biomass and carbon content of macrophytes, CO<sub>2</sub> and CH<sub>4</sub> fluxes, and the weight and carbon content of sedimental material. We also quantified the abundance of zooplankton, phytoplankton, and genes related to methanogenic and methanotrophic microorganisms. We hypothesized that increased macrophytes biomass due to HCO<sub>3</sub><sup>-</sup> addition would enhance the system's carbon stocks through the conversion of biomass carbon, and that could uptake more  $HCO_3^{-}$  and thus reduce  $CO_2$  flux. However, we postulated that this increase in biomass would have non-significant effects on CH<sub>4</sub> flux. We identified the key predictors of CH<sub>4</sub> and  $CO_2$  fluxes following  $HCO_3^-$  addition. To determine the net carbon stock of the system throughout the experiment, we integrated sedimentation, plant carbon fixation, carbon concentration in the water body into a carbon stock model.

# Methods

# **Experimental set-up**

The outdoor mesocosm experiment was conducted at Dongshan substation of Taihu Lake Ecosystem Research Station ( $31^{\circ}2'2''$ N,  $120^{\circ}25'17''$ E) in Jiangsu Province, China, from July 2023 to August 2024. The mesocosms comprised 16 opaque polyethylene barrels, each with a top diameter of 100 cm, a bottom diameter of 85 cm, and a depth of 83 cm. Prior to the experiment, *M. spicatum* was collected from Eastern Lake Taihu and pre-cultivated in a barrel for two months. Sediments collected

from Eastern Lake Taihu were spread on outdoor concrete surfaces to air dry under sunlight for 15 days, effectively minimizing the influence of benthic animals and aquatic plant seeds on the experiment. Larger debris such as dead branches, stones, and remnants of benthic fauna were manually selected and removed. The dried sediments were then placed in the barrels, and 450 L of lake water was pumped from the surface of Eastern Lake Taihu. The barrels were then left to stand for three days to allow the sediment to fully rehydrate. A mixer was used to homogenize the sediments, breaking down larger clumps into fine particles. The barrels were left undisturbed for an additional ten days to facilitate the settling of suspended sediment particles and to clarify the water. Throughout this period, floating debris and dead branches were removed from the water surface by a scoop net. Subsequently, M. spicatum of uniform height and biomass was selected from the pre-culture barrel and transplanted in a concentric circle pattern within the cleared barrels using the cutting method. Each barrel received 20 unbranched shoots of M. spicatum, with an average total biomass of  $40.24 \pm 0.36$  g (fresh weight) and an average height of 26.56±1.73 cm. Post-planting, the macrophytes were left undisturbed for one week to acclimate to their new environment. During the acclimation period, any macrophytes that failed to adapt were promptly replaced with fresh ones under the same conditions. Once the experimental system was successfully established, a beaker (7 cm in diameter, 9.9 cm in height, and 250 mL in volume) was embedded in the sediment of each barrel to collect sediments over the duration of the one-year study.

A one-way factorial experiment was designed with four levels of  $HCO_3^-$  addition (0 mmol L<sup>-1</sup>, 0.5 mmol L<sup>-1</sup>, 1 mmol L<sup>-1</sup>, and 2.5 mmol L<sup>-1</sup>, designated as "Control", "Low", "Medium", and "High", respectively). Each treatment had four replicates. HCO3<sup>-</sup> was introduced as NaHCO<sub>3</sub>. For each treatment, a precise quantity of NaHCO<sub>3</sub> powder (Sinopharm, AR) was weighed, dissolved in water, and then added to each experimental barrel, ensuring thorough mixing. Following the addition of  $HCO_3^{-}$ , the pH values of each treatment were 7.04 ± 0.13,  $7.45 \pm 0.22$ ,  $7.63 \pm 0.25$ , and  $8.06 \pm 0.34$ , respectively, which are consistent with the range typically observed for pH levels in natural waters [51]. Initially, HCO<sub>3</sub><sup>-</sup> was added once in August 2023. The addition was halted as the plant growing season drew to a close. From March 2024 to July 2024, monthly  $HCO_3^-$  additions were resumed.

The experiment was conducted over a full year, encompassing the growth and decay periods of *M. spicatum*. Throughout the duration of the experiment, water transparency in each experimental barrel was consistently sufficient to allow visibility to the bottom (approximately 70 cm). The total nitrogen concentration in the water was sustained at  $0.75 \pm 0.47$  mg L<sup>-1</sup>, and the total phosphorus concentration was kept at  $0.033 \pm 0.025$  mg L<sup>-1</sup>. These conditions were adequate to fulfill the light and nutrient requirements necessary for the normal growth of *M. spicatum*. During the experiment, water levels in the experimental barrels were kept relatively consistent. Natural rainfall was utilized to replenish water levels, and when rainfall was scarce, tap water was promptly added to prevent significant deviations in water levels. Additionally, the cylinder walls were regularly cleaned to mitigate the impact of periphyton on the experimental results.

# Water quality parameters

Water temperature (WT), pH, dissolved oxygen (DO), salinity (SAL) and specific conductance were measured by a portable multiparameter water quality meter (YSI ProQuatro, YSI Inc., USA). Raw water samples were analyzed for total nitrogen (TN), total phosphorus (TP), total organic carbon (TOC), total inorganic carbon (TIC) and alkalinity. To determine dissolved total nitrogen (DTN) and dissolved total phosphorus (DTP), water samples were filtered through GF/C filters (1.2 µm, Whatman, UK). The GF/C filters were subsequently used for chlorophyll-a (Chl-a) determination. For dissolved organic carbon (DOC) and dissolved inorganic carbon (DIC) analysis, water samples were filtered through GF/F filters (0.7 µm, Whatman, UK). TN, TP, DTN, DTP and Chl-a were analyzed with a UV spectrophotometer (UV2600, Shimadzu, Japan). Alkalinity was determined by acidbase titration [43]. TOC, TIC, DOC and DIC were analyzed by high-temperature oxidation method with a TOC analyzer (TOC-L CPH, Shimadzu, Japan). All the aforementioned water quality parameters were measured once a month.

#### Carbon sedimentation and content in sediments

Following the conclusion of the one-year experiment, the beakers embedded in the sediments were retrieved and transported to the laboratory. The beakers were left to stand for three days to facilitate the settling of suspended particles in the overlying water. The water was then carefully siphoned off, and the beakers were dried externally. The total wet weight of the sediment was recorded using an electronic balance. Subsequently, a sufficient portion of the sediment was placed in an oven at 105°C for 48 h to achieve a constant weight, after which the dry weight was measured. The dried sediment was ground using an agate mortar and passed through a 100-mesh sieve. Finally, the total carbon content of the sediments was determined using an organic elemental analyzer (Vario UNICUBE, Elementar, German).

### CH<sub>4</sub> and CO<sub>2</sub> diffusive fluxes at the water-air interface

Given that the diffusive flux is at least three orders of magnitude higher than the ebullition flux (Table S1), this study chose to focus solely on the diffusive flux. Diffusive fluxes of  $CH_4$  (FCH<sub>4</sub>) and  $CO_2$  (FCO<sub>2</sub>) were measured once a month. The measurements of FCH<sub>4</sub> and FCO<sub>2</sub> at the water–air interface were performed using the static chamber method. A portable GHG analyzer (GW-2032, Wuhan Ganwei Technology Co., Ltd., China) was utilized to measure the real-time concentration of  $CH_4$  and  $CO_2$  accumulated in the floating chamber of each experimental barrel. The floating chamber design, as described by Xun et al. [61], consisted of a plexiglass cylindrical barrel, a lid with a fan gas mixing, an air inlet, an air outlet duct, and a floating ring.

The concentration of  $CH_4$  and  $CO_2$  accumulated in the floating chamber were employed to calculate the FCH<sub>4</sub> and FCO<sub>2</sub>. The calculation was performed as follows:

$$F = \frac{(C_2 - C_1) \bullet h}{\Delta t \bullet V_m} \bullet \frac{T_1}{T_2} \bullet \frac{P_2}{P_1}$$

where *F* is the FCH<sub>4</sub> and FCO<sub>2</sub> (µmol m<sup>-2</sup> s<sup>-1</sup> or mol m<sup>-2</sup> yr<sup>-1</sup>). *C*<sub>1</sub> and *C*<sub>2</sub> (ppm) denote the gas concentration measured by portable GHG analyzer at *t*<sub>1</sub> and *t*<sub>2</sub> (s), respectively. The height of the floating chamber, *h*, is fixed at 0.4 m.  $\Delta t$  refers to the time interval between *t*<sub>2</sub> and *t*<sub>1</sub> (~15 min). *V*<sub>m</sub> is the molar volume of gas, which is taken as 22.4 L·mol<sup>-1</sup> for this experiment. *T*<sub>1</sub> is the standard temperature (273 K) and *T*<sub>2</sub> is the chamber gas temperature. *P*<sub>1</sub> is the standard atmospheric pressure (101.325 kPa), and *P*<sub>2</sub> is the chamber gas atmospheric pressure.

## Dry weight and carbon content of M. spicatum

Monthly collections of M. spicatum samples were conducted, with the exception of December 2023, January 2024, and February 2024. The dry weight of M. spicatum in each barrel was measured using a sampling survey method with a defined area of 78.5 cm<sup>2</sup>. Given that only 1.38% of the sampling area in the barrels was sampled each time, it was deemed that sampling procedure did not significantly perturb the system. The M. spicatum samples were manually harvested, with epiphytes removed by rinsing in distilled water. The fresh macrophyte samples were stored at -20°C before being dried in an oven at 50°C until a constant weight was reached. The dry weight was recorded, and the samples were ground and passed through a 100-mesh sieve. The carbon content of M. spicatum was then determined using an organic elemental analyzer (Vario UNICUBE, Elementar, German).

# Abundance of zooplankton and phytoplankton

Zooplankton and phytoplankton samples were collected once a season (in August and October in 2023 and January and April in 2024). Zooplankton samples were sampled using a 5-L plexiglass water sampler, with a total of 10 L collected in each barrel. These samples were filtered through a plankton net with a mesh size of 64  $\mu$ m and then preserved in a 4% formaldehyde solution. The two subsamples from each barrel were consolidated into a single sample. The identification of zooplankton genera was conducted using a binocular microscope. The abundance of zooplankton was calculated based on the number and size of individuals [65].

Phytoplankton samples were collected using a plexiglass water sampler to obtain 1 L of surface water, to which 10 mL of Lugol's solution was added for preservation. Following a 48-h settling period in the dark, the samples were siphoned to remove the supernatant. The residual solution was then concentrated to 30 mL and preserved. Identification and enumeration were performed using an optical microscope. The two subsamples from each barrel were pooled into a single sample. The abundance of phytoplankton was calculated by converting the cell density of various genera to volume [19].

# Abundance of methanogens and methanotrophs in sediments

Microorganism samples were collected from surface sediments on a quarterly basis in alignment with the zooplankton and phytoplankton sampling schedule, specifically in August and October of 2023, and January and April of 2024. The four replicated samples from each treatment were homogenized to form a single representative sample for that treatment, and then stored in ziplock bags at -20°C. High-throughput sequencing of sediments was conducted to elucidate the specific gene abundance. Quantitative real-time polymerase chain reaction (qPCR) fluorescence quantification was employed to selectively amplify and quantify specific genes (mcrA for methanogenic archaea and *pmoA* for methanotrophic bacteria) [24, 25]. The primers employed for the *mcrA* gene were a forward primer MLf (5'- GGTGGTGTMGGATTCACA CARTAYGCWACAGC-3') and a reverse primer MLr (5'-TTCATTGCRTAGTTWGGRTAGTT-3') [30]. For the pmoA gene, the primer set included A189f (5'-GGNGAC TGGGACTTCTGG-3') and Mb661r (5'-CCGGMGCA ACGTCYTTACC-3') [7, 18]. For PCR amplification, the cycling conditions were as follows: an initial denaturation step at 95°C for 2 min, 25 cycles (denaturation at 95°C for 30 s, annealing at 55°C for 30 s, and extension at 72°C for 30 s), followed by a final extension at 72°C for 5 min. For the quantitative analysis of methanogens and methanotrophs, the TB Green fluorescence quantitative PCR kit (Takara, Japan) was applied in conjunction with the aforementioned primers and DNA templates. Quantitative analysis was performed using the QuantiFluor<sup>™</sup>ST Blue Fluorescence Quantitation System on a NextSeq<sup>™</sup> 2000 platform at Personal Biotechnology Co., Ltd (Shanghai, China). Data were analysed using the online platform Genescloud (www.genescloud.cn).

# Calculations of carbon stocks in the macrophyte-dominated system

In our study, the primary carbon stocks within the macrophyte-dominated system were categorized into three distinct compartments: macrophyte carbon stock (MCS), water column carbon stock (WCS), and sedimentation carbon stock (SCS). The total carbon stock (TCS) was composed of MCS, WCS, and SCS, with the amount of artificially added carbon (AC) deducted. In a system where macrophytes are dominant, the carbon storage capacity of microorganisms and plankton is significantly lower compared to that of macrophytes. Additionally, the sediments within the system have undergone pre-exposure, and the presence of large benthic animals is minimal. Consequently, the carbon contribution from these organisms was excluded from our study's analysis. The calculation formulas are as follows:

 $MCS = Weight \times Carbon content$ 

$$WCS = (TOC + TIC) \times V$$

$$SCS = \frac{SC \times DSW}{A_{heaker} \times t}$$

$$TCS = MCS + WCS + SCS - AC$$

where MCS (g C) represents the macrophyte carbon stock at the end of the experiment. Weight (g) is the dry weight of all macrophytes in each barrel at the end of the experiment. WCS (g C) represents the water column carbon stock at the end of the experiment. TOC and TIC (g  $C L^{-1}$ ) denote their concentrations in water at the end of the experiment, respectively. V(L) is the volume of water in the barrel (450 L). SCS (g C) refers to the sedimentation carbon stock at the end of the experiment. SC (%) represents the total carbon content of sediments, with its determination method detailed in Sect. 2.3. DSW (g) is the dry sediment weight collected in the beaker. A beaker represents the bottom area of the beaker (38.47 cm<sup>-2</sup>). t (yr) is the duration for which the beaker is placed. TCS (g C) represents the total column carbon stock at the end of the experiment. AC (g C) represents the added carbon.

### Statistics

All statistical analyses were performed using R v4.4.2 (R Core Team, 2024). We evaluated the effects of varying  $HCO_3^-$  concentrations on each observed variable and GHG diffusive fluxes. Prior to analyzing the response to the addition of  $HCO_3^-$ , normality and homogeneity of variance tests were conducted for each variable. The outcomes of these tests dictated the selection of parametric (ANOVA) or non-parametric (Kruskal–Wallis test) methods, followed by the corresponding post-hoc tests. The statistical analyses employed included oneway ANOVA, Welch's ANOVA, and Kruskal–Wallis test, with their respective post-hoc tests being Tukey's HSD test, Games-Howell test, and Dunn test.

To explore the of each observed variable on GHG diffusive fluxes, we utilized the *gamm* and *gam* functions from the "mgcv" package to construct generalized additive mixed models (GAMM) and generalized additive models (GAM), respectively. These models were used to examine the response curves of GHG diffusive fluxes to each observed variable [56]. To minimize the impact of scale differences among predictors and improve the stability and explanatory capacity of the model, the data underwent standardization prior to the construction the GAMM or GAM models. All observed variables were incorporated as fixed effects, with FCH<sub>4</sub> and FCO<sub>2</sub> as the response variables. In GAMM models, different parallel replicates were designated as random effects to account for the intrinsic correlation among them.

The variables exhibiting a significant response to  $HCO_3^-$  addition, along with those identified in GAMM or GAM models as having a significant correlation with GHG diffusive flux, were selected for further analysis. These variables were then subjected to Recursive Feature Elimination (RFE) with Random Forests (RF) to determine the key predictors for FCO<sub>2</sub> and FCH<sub>4</sub>. RF employed the mean decrease in accuracy (MDA) to calculate the feature importance from the training model of RF-RFE. Functions from the "randomForest" and "rfPermute" packages were employed to construct the RF model and extract variable importance and significance. Variables with the least contribution to model prediction were eliminated, thereby refining the final input feature set through RFE. The variables optimized by RF-RFE were utilized in a Partial Least Squares Path Model (PLS-PM) for causal path analysis, aiming to elucidate the effect of  $HCO_3^-$  addition on  $FCH_4$  and  $FCO_2$ , as well as to determine the direct and indirect effects of each feature variable on FCH<sub>4</sub> and FCO<sub>2</sub>. The plspm function from the "plspm" package was employed to construct the PLS-PM model.

# Results

# Responses of FCH<sub>4</sub> and FCO<sub>2</sub> to HCO<sub>3</sub><sup>-</sup> addition

average The annual  $FCH_4$ values were  $m^{-2}$  $\mathrm{yr}^{-1}$  $0.11 \pm 0.12$ mol in the control,  $0.07\pm0.05\ mol\ m^{-2}\ yr^{-1}$  in the low-level  $HCO_3^{-}$  treatment,  $0.07 \pm 0.07$  mol m<sup>-2</sup> yr<sup>-1</sup> in the medium-level  $HCO_3^{-}$  treatment, and  $0.07\pm0.09~mol~m^{-2}~yr^{-1}$  in the high-level HCO3<sup>-</sup> treatment, respectively. No significant differences were observed between the different treatments (Fig. 1a). The temporal variation in CH<sub>4</sub> diffusion flux followed a similar pattern across all treatments (Figure S1). In contrast, the  $FCO_2$ showed a significant response to the different levels of  $HCO_3^-$  addition (Fig. 1b). The annual average  $FCO_2$ values were  $-3.48 \pm 7.60$  mol m<sup>-2</sup> yr<sup>-1</sup> in the control, -6.78  $\pm\,5.87$  mol  $\mathrm{m^{-2}\,yr^{-1}}$  under low  $\mathrm{HCO_{3}^{-}}$  addition conditions, -7.15  $\pm$  8.68 mol m<sup>-2</sup> yr<sup>-1</sup> under medium HCO<sub>3</sub><sup>-</sup> addition conditions, and -14.04  $\pm$  14.39 mol m  $^{-2}$  yr  $^{-1}$ under high HCO<sub>3</sub><sup>-</sup> addition conditions, respectively (Figure S2).

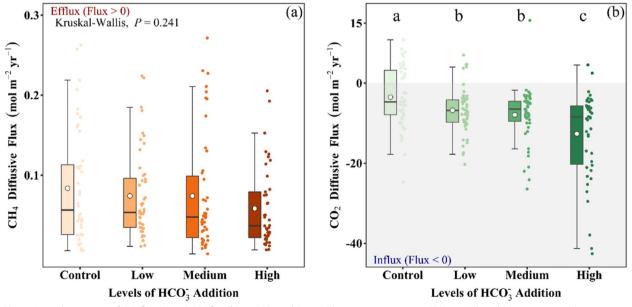
# Responses of other biotic and abiotic variables to $\mathrm{HCO_3}^-$ addition

In the treatment with high-level  $\text{HCO}_3^-$  addition, TP, DTP, Chl-*a*, TOC, and DOC concentrations were significantly elevated compared to the other three treatments (*P*<0.05), with no significant differences among the latter three (Table S2). The average values of alkalinity, pH, TIC, and DIC also increased with increasing

 $HCO_3^-$  levels and exhibited significant differences across the different treatments. The addition of  $HCO_3^-$  had an influence on the community composition of zooplankton and phytoplankton (Figure S3 and S4); nonetheless, it did not significantly affect their abundance (Table S2). The addition of  $HCO_3^-$  led to an increase in the dry weight of *M. spicatum*, with higher levels of  $HCO_3^-$  corresponding to significantly great dry weight accumulation (*P* < 0.05). In the treatment with high-level  $HCO_3^-$  addition, the annual average dry weight was approximately 2.5 times higher than that in the control. However, carbon content in *M. spicatum* did not show significant responses to  $HCO_3^-$  addition.

In the sediments, the absolute abundances of *mcrA* (methanogenic archaea) and *pmoA* (methanotrophic bacteria) genes generally decreased with increasing  $HCO_3^-$  addition (Table S2). The absolute abundance of *mcrA* was highest in the control and lowest in the high-level  $HCO_3^-$  treatment. A general decline of the abundance of *pmoA* was observed alongside increasing  $HCO_3^-$  addition. Despite these downward trends, there were no significant differences in absolute abundance across different treatments.

**Correlations between various variables and FCH<sub>4</sub> and FCO<sub>2</sub>** The GAMM and GAM models revealed significant correlations between FCH<sub>4</sub> and a range of variables, including water temperature, TP, TOC, DOC, dry weight of *M. spicatum*, abundance of zooplankton and phytoplankton,



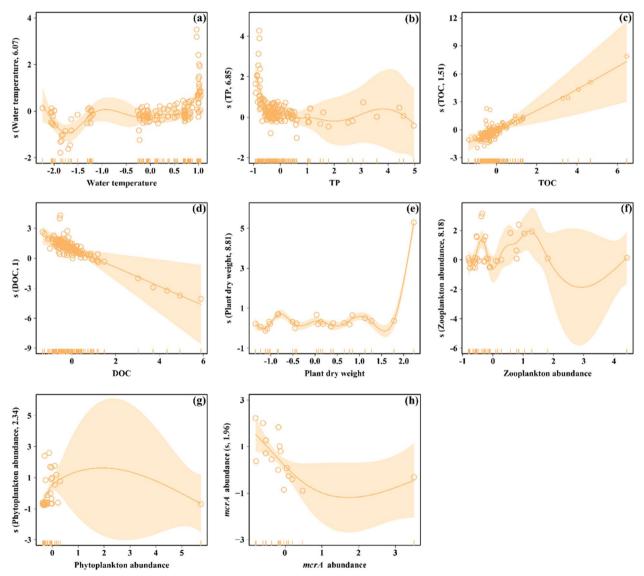
**Fig. 1** Annual responses of  $CH_4$  flux (**a**) and  $CO_2$  flux (**b**) to  $HCO_3^-$  addition. Values represent mean  $\pm$  SD (n = 4). The whiskers represent the 95% confidence intervals. The categories Control, Low, Medium, and High correspond to  $HCO_3^-$  addition levels of 0, 0.5, 1, 2.5 mmol·L.<sup>-1</sup>, respectively. Significant differences between treatments are indicated different letters, with P < 0.05

and absolute abundance of *mcrA* (Fig. 2). Notably, the correlation between DOC and FCH<sub>4</sub> was nearly linear (as indicated by a small degree of freedom, df=1), whereas the relationships with the other parameters were more complex and nonlinear (Fig. 2 and Table S3). Regarding FCO<sub>2</sub>, significant correlations were observed with water temperature, pH, DO, conductance, TN, TP, DTN, alkalinity, dry weight and carbon content of *M. spicatum*, and number of phytoplankton (Fig. 3). Among these, water temperature, pH, DO, conductance, DTN, alkalinity, and

dry weight of *M. spicatum* exhibited an overall negative influence on  $FCO_2$ , indicating that the reductions in  $FCO_2$  were associated with these parameters.

# Key predictors of FCH<sub>4</sub> and FCO<sub>2</sub>

According to the RF-RFE analysis, key predictors for FCH<sub>4</sub> were identified as DOC, conductance, TIC, abundance of phytoplankton, DIC, salinity, abundance of *mcrA*, and water temperature, with DOC emerging as the sole statistically significant predictor (Fig. 4). For FCO<sub>2</sub>,



**Fig. 2** Generalized Additive Mixed Models (GAMM) and Generalized Additive Models (GAM) results for CH<sub>4</sub> flux. Scatter points represent the residuals between the observed values and the model predictions. The shaded regions encompass the 95% confidence interval. The "rug" along the x-axis and y-axis displays the density of observations. The number adjacent to each y-axis label indicates the effective degrees of freedom for the plotted term. The *y*-axis values indicate *x*-axis covariate effects on deviations from the mean prediction (continuous line). This line represents an estimate of the smooth function of partial residuals, indicating the *x*-axis covariate effects on the measured trait. When the effective degree of freedom equal one, the continuous line is a straight line, indicating a linear effect of the *x*-variable

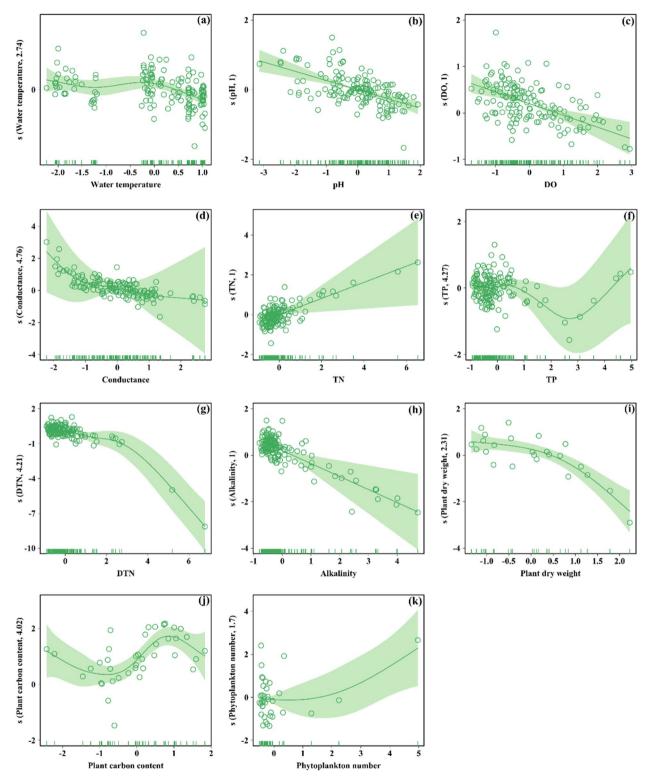
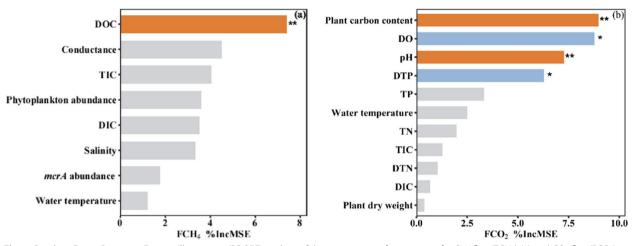


Fig. 3 GAMM and GAM results for CO<sub>2</sub> flux. Scatter points represent the residuals between the observed values and the model predictions



**Fig. 4** Random Forest Recursive Feature Elimination (RF-RFE) analysis of the importance of parameters for  $CH_4$  flux (FCH<sub>4</sub>) (a) and  $CO_2$  flux (FCO<sub>2</sub>) (b). The x-axis represents the relative feature importance of variables, as calculated by the mean decrease in accuracy (MDA) method, with \*P < 0.05 and \*\*P < 0.01 in RF analysis

eleven key predictors were identified and the significant predictors were carbon content of *M. spicatum*, DO, pH, and DTP (Fig. 4). The significant predictors from the RF-RFE analysis were incorporated into the PLS-PM analysis as direct influencing variables for FCH<sub>4</sub> and FCO<sub>2</sub>, whereas the non-significant were used as indirect influencing variables. The PLS-PM analysis results suggest that the addition of HCO<sub>3</sub><sup>-</sup> exerts its regulatory effect on FCH<sub>4</sub> primarily through its influence on DOC, making DOC the predominant direct factor (Fig. 5a and c).

In the PLS-PM model elucidating  $FCO_2$ , pH was identified as the predominant direct factor, contributing to 87.69% of the direct influence on  $FCO_2$ , with a minor indirect influence of 12.31% (Fig. 5b). Similarly, phosphorus and DO had both direct and indirect effects on  $FCO_2$ , with the direct influence being more pronounced (Fig. 5d). In contrast, the carbon content of *M. spicatum* exerted a direct influence on  $FCO_2$  without any discernible indirect impact (Fig. 5d). Post the introduction of  $HCO_3^-$ , the combined effect of these factors led to a significant reduction in  $FCO_2$ .

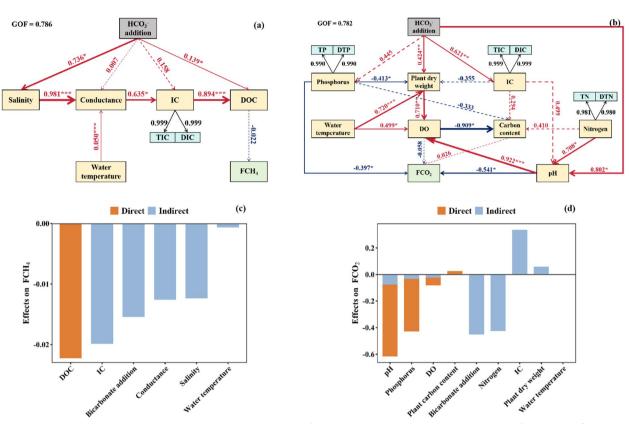
# Responses of carbon stock to HCO<sub>3</sub><sup>-</sup> addition

The addition of  $HCO_3^-$  resulted in a significant increase in the water column carbon stock (WCS) and a minor impact on macrophyte carbon stock (MCS) and total carbon stock (TCS), with no effect on sedimentation carbon stock (SCS) (Table S4). With the increase in  $HCO_3^$ addition levels, the proportions of WCS and MCS in TCS increased, while the proportion of SCS gradually decreased (Fig. 6). The response of WCS to  $HCO_3^-$  addition was consistent with the changes in TIC and DIC (Table S2). At the end of the experiment, the WCS values for each treatment were  $6.67 \pm 0.71$  g C,  $10.41 \pm 2.94$  g C,  $13.75 \pm 0.23$  g C, and  $31.18 \pm 5.84$  g C, respectively, with significant differences noted between the high-level HCO<sub>3</sub><sup>-</sup> addition and the control (*P*<0.05) (Table S4). Despite an increase in the mean value and proportion of MCS with higher HCO<sub>3</sub><sup>-</sup> addition, no significant differences were observed among the treatments at the end of the experiment. Additionally, the addition of HCO<sub>3</sub><sup>-</sup> had a negative impact on the proportion of SCS and even led to a slight decrease in sediment carbon accumulation. And the addition of HCO<sub>3</sub><sup>-</sup> enhanced the capacity of the total carbon stock of the entire system (Fig. 6 and Table S4).

#### Discussion

# Effects of elevated DIC input on CO<sub>2</sub> and CH<sub>4</sub> fluxes

Rapid carbon cycling in shallow lakes, ponds, and wetlands, compared to deep lakes, is largely attributed to their vulnerability to human activities such as agriculture and urbanization. These activities result in significant inputs of DIC and nutrients within these aquatic ecosystems [20, 66]. Submerged macrophytes have adapted to conditions of low CO<sub>2</sub> availability in aquatic environments by developing physiological structures capable of utilizing DIC and enzymatic systems that utilize  $HCO_3^{-}$ . They can also catalyze the conversion between HCO<sub>3</sub><sup>-</sup> and CO<sub>2</sub> through carbonic anhydrase, thus maintaining high photosynthetic rates [29]. Upon exposure to light, the activity of H-ATPase increases, lowering the pH in the epidermis and consequently reducing the  $HCO_3^{-}/$ CO<sub>2</sub> ratio, which allows for the absorption and utilization of more  $CO_2$  [50]. This, in turn, decreases the  $CO_2$ emissions in macrophyte-dominated systems. While

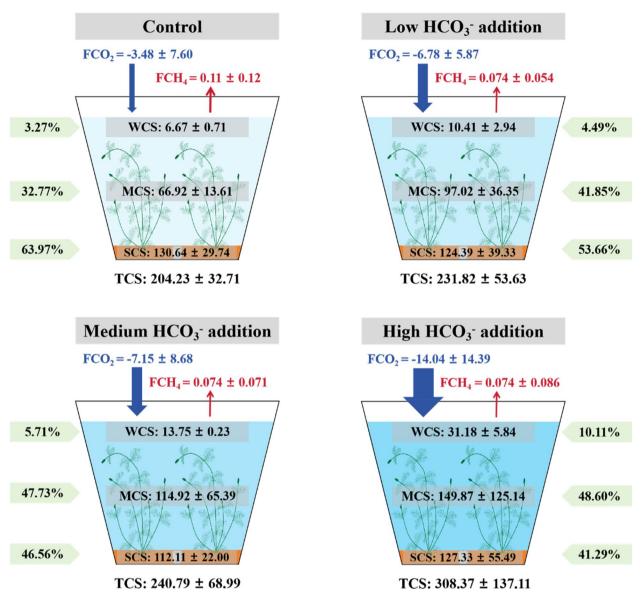


**Fig. 5** Partial Least Squares Path Model (PLS-PM) elucidates the impact of  $HCO_3^-$  on  $FCH_4$  (**a**) and  $FCO_2$  (**b**). The thickness of the arrows reflects the magnitude of the path coefficients. Solid arrows indicate significant paths, whereas dashed arrows represent non-significant paths. Red and blue lines and numbers correspond to positive and negative paths and their respective strengths. The black numbers signify the loadings of indicators (blue rectangles) onto their corresponding latent variables. "\*" (P < 0.05) and "\*\*" (P < 0.01) indicate that the path coefficients are significant at the 95% confidence level. Standardized direct and indirect effects on  $FCH_4$  (**c**) and  $FCO_2$  (**d**) were derived from the PLS-PM outcomes

acknowledging that an increase in the biomass of macrophytes leads to enhanced accumulation of organic matter, heightened microbial activity, and improved organic matter decomposition efficiency, consequently affecting respiratory activity and  $CO_2$  emissions [53, 64], our findings demonstrate that submerged vegetation substantially diminishes  $CO_2$  fluxes at the water–air interface.

Notably, elevated  $\text{HCO}_3^-$  concentrations correlate with an increase in macrophyte biomass and a corresponding decrease in CO<sub>2</sub> fluxes (Fig. 1 and Table S2). Shortterm studies have indicated that high-density submerged macrophytes exhibit elevated CO<sub>2</sub> concentrations and emissions within microcosms [48]. Other studies have reported that the impact of submerged macrophytes on lake CO<sub>2</sub> emissions is not significant over a threemonth period [14, 16]. In contrast, our experimental period spanned one year, encompassing both the growth and decomposition phases of the macrophytes, indicating that systems dominated by submerged macrophytes are beneficial for reducing the system's CO<sub>2</sub> emissions. This finding aligns with previous research conducted in a macrophyte-rich lake [58]. The discrepancies observed in the aforementioned studies could be attributed to the multifaceted impact of submerged plants on  $CO_2$  emissions. Factors such as varying environmental conditions, species and density of submerged plants, and notably, the study duration, are likely to have influenced the outcomes.

Our study revealed that pH, DO, and phosphorus content in the water were negatively correlated with  $CO_2$  fluxes (Fig. 5b), which is consistent with previous findings [45]. As water pH rises from 7.0 to 8.5, the ratio of  $HCO_3^-$  to  $CO_2$  concentration increases markedly, from 4 to 140 times [6]. Consequently, submerged macrophytes preferentially utilize  $HCO_3^-$  under these conditions. In lakes receiving high DIC inputs,  $CO_2$  emissions are greatly influenced by pH [49]. Our findings confirm that pH is a primary factor affecting  $CO_2$  fluxes with  $HCO_3^-$  addition (Fig. 5d). Post  $HCO_3^-$  addition, the water pH increased,  $CO_2$  partial pressure decreased, leading to a significant reduction in  $FCO_2$  (Fig. 3). Phosphorus content, particularly DTP,



**Fig. 6** Summary of carbon stock and  $CO_2$  and  $CH_4$  fluxes (mean ± SD, n = 4) under different levels of  $HCO_3^-$  addition in macrophyte-dominated systems, with WCS, MCS, SCS and TCS at the end of the experiment in g C, and annual average  $FCH_4$  and  $FCO_2$  in mol  $m^{-2}$  yr<sup>-1</sup>. WCS means water column carbon stock. MCS refers to macrophyte carbon stock. SCS represents sedimentation carbon stock. TCS represents total carbon stock

also significantly impact FCO<sub>2</sub> (Fig. 4b). Phosphorus is an essential nutrient for submerged macrophytes, enhancing CO<sub>2</sub> uptake by boosting primary production [11]. With increased HCO<sub>3</sub><sup>-</sup> addition, our study observed a rise in dry weight of macrophyte (Table S2). This could lead to enhanced consumption of dissolved CO<sub>2</sub> and production of DO within aquatic systems, as supported by previous studies [10, 40], thereby reducing CO<sub>2</sub> fluxes and establishing a negative correlation between CO<sub>2</sub> fluxes and the concentrations of DO and phosphorus.  $CH_4$  in aquatic systems such as lakes and wetlands is primarily produced through the anaerobic decomposition of organic matter by methanogenic archaea in sediments [44]. Additionally, some researchers have indicated that submerged plants can produce methyl compounds that are converted into  $CH_4$  under the influence of reactive oxygen species, and endophytic or epiphytic archaea, algae, cyanobacteria, and proteobacteria on plants can also produce methane in aerobic environments [4, 5, 17, 38, 44]. In our study, as the concentration of  $HCO_3^-$  increased, the absolute abundance of the *mcrA* gene (methanogenic archaea) in sediments showed a decreasing but non- significant trend. Nonetheless, the abundance of the *pmoA* gene (methanotrophic bacteria) also demonstrated a gradual but non-significant decline (Table S2). Concurrently,  $CH_4$  fluxes also exhibited a downward trend, albeit not significantly (Fig. 1). This finding reflects that submerged plant-dominated systems have a slight mitigating effect on  $CH_4$  emissions; however, this effect is not as pronounced as the reduction in  $CO_2$  emissions.

Oxygen secretion from plant leaves and roots may lead to an increase in DO concentration in both water columns and sediments [40]. In this study, DO concentrations showed a gradual increase with the addition of HCO<sub>3</sub><sup>-</sup>, however, this trend was not significant (Table S2), and thus no significant correlation was observed between DO and CH<sub>4</sub> fluxes (Fig. 4). Following the addition of HCO<sub>3</sub><sup>-</sup>, DIC can be partially converted into DOC by organisms, and an increase in DOC concentration in water has a direct negative impact on CH<sub>4</sub> emissions (Fig. 5a and c). Submerged macrophytes contain a higher proportion of refractory components, which are prone to forming a refractory carbon pool [54]. One study demonstrated that the annual decomposition rate of submerged plant biomass was less than 50%, suggesting that more than half of the biomass of submerged macrophytes is involved in the carbon sequestration process on an annual basis [26]. Therefore, the increasing DOC concentrations in this study with the gradual addition of HCO<sub>3</sub><sup>-</sup> indicate that DOC is not rapidly decomposed and converted into CH<sub>4</sub>. Collectively, the results of the one-year experiment showed that the CH<sub>4</sub> emissions from systems dominated by macrophytes were slightly but not significantly inhibited (Fig. 1a). Consequently, even though submerged macrophytes might directly or indirectly produce CH<sub>4</sub>, the CH<sub>4</sub> fluxes of the systems did not significantly increase, and even showed a slight decrease. Although some studies have found that CH<sub>4</sub> flux did not significantly change after the harvest of macrophytes [14, 16], our one-year investigation conducted in Lake Xuanwu (Nanjing, China) revealed that in the ecologically restored parts dominated by submerged macrophytes, both CH<sub>4</sub> flux and concentration in the water body were significantly lower than in the unrestored eutrophic parts dominated by phytoplankton [31].

# Effects of elevated DIC input on water carbon stock (WCS)

Water carbon stock typically encompasses both inorganic and organic carbon forms. DIC serves as a pivotal reactant and product in the processes of DOC formation and degradation [32, 55]. POC constitutes a minor fraction of the total carbon in aquatic systems, with over 90% of the organic carbon existing in the form of DOC, particularly in marine environments [47]. Particulate organic carbon (POC) is also decomposed by microorganisms into DOC, a significant portion of which is subsequently transformed into refractory components [15]. In our study, the addition of  $HCO_3^-$  led to an increase in the organic carbon stock through enhancing the conversion of DIC, thereby expanding the overall capacity of the WCS. Additionally, the elevation of pH due to  $HCO_3^-$  addition facilitated the complexation of DOC more with metal ions such as calcium and magnesium, reducing the photodegradation and mineralization rates of DOC [33], and consequently, the capacity of the organic carbon stock was expanded.

After the initial single addition of  $HCO_3^-$ , WCS exhibited an increase in the subsequent month (Figure S5). However, this temporary increase was hard to sustain and tended to decline over time. In contrast, with monthly additions of  $HCO_3^-$ , WCS not only maintained a high capacity but also showed a tendency of further expansion over time (Figure S5). Macrophytes continuously consumed  $HCO_3^-$  during primary production, accelerating the transformation of DIC into DOC [9]. In natural lakes, where watershed DIC typically enters in a continuous manner, the WCS was expected to increase in systems dominated by submerged macrophytes.

# Effects of elevated DIC input on macrophyte carbon stock (MCS)

The increase in DIC can accelerate the relative growth rate of some submerged macrophytes and promote more branching [9], enhance biomass [28], and alleviate carbon competition with periphyton [23]. Consequently, in our experiment, the positive response of the MCS on HCO<sub>3</sub><sup>-</sup> was observed (Table S2). The increase in DIC concentration led to an increase in MCS capacity, allowing more organic carbon to be stored within MCS and strengthening the carbon sequestration capacity of the entire system (Fig. 6). The carbon fixation capacity of aquatic macrophytes is estimated to be approximately  $2.04 \times 10^{18}$  g yr<sup>-1</sup>, with a carbon storage amount reaching  $1.17 \times 10^{17}$  g yr<sup>-1</sup> [35]. In Lake Baoan (Wuhan, China), the dominant macrophyte species Potamogeton crispus L. annually sequesters approximately 288 g m<sup>-2</sup> of carbon [36]. Research indicates that the annual decomposition rate of submerged plant biomass is less than 50%, suggesting that over half of the submerged plant biomass contributes to interannual carbon storage [26]. Consequently, submerged macrophytes represent a substantial carbon reservoir within aquatic ecosystems [36]. In conclusion, in aquatic ecosystems characterized by submerged macrophytes, these plants, via physical, chemical, and biological processes, prolong the carbon turnover time, thereby enhancing carbon sequestration.

# Effects of elevated DIC input on sedimentation carbon stock (SCS)

Research indicates that when the total primary productivity carbon content in lakes exceeds 25 g C  $m^{-2}$  yr<sup>-1</sup>, the water body transitions from a carbon sink to a source, releasing the sequestered  $CO_2$  back into the atmosphere. Consequently, in eutrophic lakes, little carbon is effectively buried [21]. As previously discussed, the addition of HCO<sub>3</sub><sup>-</sup> directly expanded the WCS and further increases the MCS through photosynthesis. DOC from the WCS, once converted to POC, and a portion of the debris or remains from the MCS and other organisms are eventually deposited and accumulated, thereby inputting into the SCS [8]. In our mesocosm experiment, the annual sedimentation rate (112–131 g C m<sup>-2</sup>·yr<sup>-1</sup>) (Fig. 6) fell within the range of sedimentation rates measured in global lake studies  $(4-400 \text{ g C m}^{-2} \cdot \text{yr}^{-1}, [46, 52].$ Sedimentation rates and amounts can be influenced by factors such as temperature, carbon composition, and water trophic status [3, 13].

Notably, our study did not observe significant changes in the sedimentation rate due to HCO<sub>3</sub><sup>-</sup> addition at the end of the experiment (Fig. 6). This observation aligns with a previous study attributing such phenomenon to carbon loss resulting from the gradient diffusion of DOC along the continuum of the sediment-pore water-overlying water [27]. More importantly,  $HCO_3^-$  addition may have promoted the metabolism of sedimentary carbon, as DIC was positively correlated with sediment respiration rate, implying that microbial carbon metabolic activity in sediments is a significant inorganic carbon source in the water column [60]. In this study, the concentrations of both organic and inorganic carbon in the water column increased with the addition of DIC, also suggesting that carbon decomposition in sediments becomes an important source of both organic and inorganic carbon in the water column (Table S2). Dense macrophyte communities can reduce sediment resuspension [22], which is not only beneficial for maintaining the ecological restoration effects of eutrophic shallow lakes but also for stabilizing the system's sedimentary carbon pool.

# **Conclusions and implications**

Our findings demonstrated that increased  $HCO_3^-$  addition reduced  $CO_2$  fluxes in *M. spicatum*-dominated systems. Both direct and indirect effects on  $FCO_2$  were observed due to variations in pH, phosphorus, and DO. The carbon content of *M. spicatum* exerted a direct influence on  $CO_2$  fluxes without any discernible indirect impact. Unlike changes in  $CO_2$  fluxes, the addition of  $HCO_3^-$  had no significant effect on  $CH_4$  fluxes. There were no significant differences in absolute abundance of *mcrA* and *pmoA* genes across different  $HCO_3^-$  addition

treatments. The regulatory effect of  $\text{HCO}_3^-$  addition on  $\text{CH}_4$  fluxes was primarily through its influence on DOC. Enhanced DIC inputs expanded the capacity for water carbon stocks and macrophyte carbon stocks. While  $\text{CH}_4$  fluxes and carbon sedimentation stocks were not significantly impacted, the overall carbon sink function of macrophyte-dominated systems was enhanced.

As climate warming intensifies, the reduction of greenhouse gas emissions becomes increasingly urgent, and the utilization of natural ecosystems for carbon storage has emerged as a focal point of global concern. Our findings in this study underscore the importance of submerged macrophyte-dominated aquatic ecosystems in mitigating CO<sub>2</sub> and CH<sub>4</sub> emissions and enhancing system carbon stocks. Our data reveal that both CO<sub>2</sub> and CH4 fluxes in the macrophyte-dominated regions of Lake Xuanwu are significantly lower than in phytoplanktondominated areas, as previously reported [31]. The ongoing restoration efforts focused on submerged plants in shallow lakes are crucial in this context, especially given the continuous influx of DIC and nutrients from the watershed into the lakes. These initiatives not only facilitate the restoration of clear water states in lakes but also augment the lakes' carbon storage capacity. Therefore, adopting effective measures to preserve and enhance the carbon sequestration potential of ecosystems, such as lakes and wetlands, is imperative for bolstering the global carbon sink and alleviating the impacts of climate warming.

#### Supplementary Information

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Supplementary Material 1.

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## Author's contributions

KYL, YLS, BHG, and FD made substantial contributions to the conception and design of the work. FD, AA, WJQ, and TQ completed the experiment and made the acquisition, analysis, and interpretation of data. FD drafted the work and YLS substantively revised it.

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#### Data availability

The datasets generated during this study are available from the corresponding author upon reasonable request.

### Declarations

**Ethics approval and consent to participate** Not applicable.

not applicable.

# Consent for publication

We confirm that all authors have reviewed and approved the final version of this manuscript and consent to its publication.

#### **Competing interests**

The authors declare no competing interests.

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